Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

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Key indicators

Single-crystal X-ray study T = 298 KMean σ (P–O) = 0.003 Å R factor = 0.025 wR factor = 0.062 Data-to-parameter ratio = 14.3

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

A new polymorph of CsPr(PO₃)₄

A new polymorph of caesium praseodymium polyphosphate, $CsPr(PO_3)_4$, features PrO_8 polyhedra sharing O atoms with PO_4 groups, yielding a three-dimensional framework. The latter delimits intersecting channels in which the Cs ions are located.

Received 9 March 2007 Accepted 3 April 2007

Comment

Rare earth phosphate materials have been extensively investigated due to their efficient luminescent and scintillation properties (Dornauf & Heber, 1979; Mazurak *et al.*, 1984; Wolinski *et al.*, 1990; Horchani *et al.*, 2002, 2003; Horchani-Naifer *et al.*, 2006; Makhov *et al.*, 2002). Many rare earth polyphosphates have been reported in the literature (Amami *et al.*, 2005; Durif, 1995). These phosphates are obtained by the flux method after study of the phase-equilibrium diagrams of the $M^{I}PO_{3}$ - $Ln(PO_{3})_{3}$ systems (M^{I} = alkali metal; Ln = rare earth element) (Férid *et al.*, 1998). This work was carried out within the framework of a systematic investigation of the crystal structures and luminescent properties of double polyphosphates of the type $M^{I}Ln(PO_{3})_{4}$ (Férid, 2006). This paper deals with the structure of a new polymorph of CsPr(PO₃)₄ [previous polymorph: Palkina *et al.* (1978)].



Figure 1

A segment of the polyphosphate chain with PO₄ tetrahedra; anisotropic displacement parameters are drawn at the 50% probability level. [Symmetry codes: (i) $x + \frac{1}{2}, -y + \frac{3}{2}, z + \frac{1}{2}$; (ii) x, y + 1, z; (iii) -x, -y + 2, -z + 1; (iv) $-x + \frac{1}{2}, y - \frac{1}{2}, -z + \frac{1}{2}$; (v) $x - \frac{1}{2}, -y + \frac{3}{2}, z - \frac{1}{2}$; (vi) $-x - \frac{1}{2}, y + \frac{1}{2}, -z + \frac{1}{2}$; (vii) -x, -y + 2, -z; (viii) $x - \frac{1}{2}, -y + \frac{5}{2}, z - \frac{1}{2}$]

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inorganic papers

CsPr(PO₃)₄ is isostructural with TlNd(PO₃)₄ (Palkina *et al.*, 1977). The central feature is a Pr³⁺ ion, which is coordinated by PO₄ groups, yielding a PrO₈ dodecahedron with Pr–O distances ranging from 2.394 (3) to 2.526 (3) Å. The PrO₈ dodecahedron shares all its O atoms with the corners of neighbouring PO₄ tetrahedra, whereas 11 O atoms constitute the coordination of the Cs atom, forming CsO₁₁ polyhedra with Cs–O distances ranging from 3.092 (3) to 3.568 (3) Å. The P–O distances of the PO₄ tetrahedra may be divided into linking or bridging P–O_i distances [1.589 (3)–1.612 (3) Å] and terminal P–O_j distances [1.480 (3)–1.492 (3) Å] (Fig. 1). The O–P–O angles are in the range 98.33 (15)–121.18 (18)°, which is in good agreement with those usually observed in polyphosphate anions (Durif, 1995).

Experimental

All reagents were used as purchased (Fluka, 99.9%). Single crystals of CsPr(PO₃)₄ were prepared by the flux method. At room temperature, Cs₂CO₃ (5 g) and Pr₆O₁₁ (0.4 g) were added slowly to phosphoric acid, H₃PO₄ (85%; 12 ml). The mixture was then heated slowly to 573 K and kept at this temperature for 7 d. Green crystals of (I) were separated from the excess phosphoric acid by washing the product in boiling water.

Crystal data

CsPr(PO₃)₄ $M_r = 589.70$ Monoclinic, $P2_1/n$ a = 10.4938 (3) Å b = 9.0744 (2) Å c = 11.2525 (5) Å $\beta = 106.414$ (2)°

Data collection

Enraf-Nonius CAD-4 diffractometer Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996) $T_{\rm min} = 0.314, T_{\rm max} = 0.320$ 3875 measured reflections

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.025$ $wR(F^2) = 0.062$ S = 1.162339 reflections T = 298 (2) K0.14 × 0.13 × 0.13 mm

V = 1027.85 (6) Å³

Mo $K\alpha$ radiation

 $\mu = 8.92 \text{ mm}^-$

164 parameters

 $\Delta \rho_{\rm max} = 1.39 \text{ e} \text{ Å}^{-3}$

 $\Delta \rho_{\rm min} = -2.29 \text{ e } \text{\AA}^{-3}$

Z = 4

2339 independent reflections 2211 reflections with $I > 2\sigma(I)$ $R_{int} = 0.018$ 2 standard reflections every 150 reflections intensity decay: none

The highest peak and the deepest hole in the residual electron density are located 0.01 Å and 0.78 Å, respectively, from Pr.

Data collection: *CAD-4 EXPRESS* (Enraf–Nonius, 1994); cell refinement: *CAD-4 EXPRESS*; data reduction: *XCAD4* (Harms & Wocadlo, 1995); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *DIAMOND* (Brandenburg, 2001); software used to prepare material for publication: *SHELXL97*.



Figure 2

A projection of $CsPr(PO_3)_4$ along the *a* axis, showing the arrangement of the PrO₈ polyhedra and PO₄ tetrahedra.

References

- Amami, J., Férid, M. & Trabelsi-Ayedi, M. (2005). Mater. Res. Bull. 40, 2144–2152.
- Brandenburg, K. (2001). *DIAMOND*. Release 2.1e. Crystal Impact GbR, Bonn, Germany.
- Dornauf, H. & Heber, J. (1979). J. Lumin. 20, 271-281.
- Durif, A. (1995). Crystal Chemistry of Condensed Phosphates. New York: Plenum Press.
- Enraf-Nonius (1994). CAD-4 EXPRESS. Version 5.1/1.2. Enraf-Nonius, Delft, The Netherlands.

Férid, M. (2006). Etude des propriétés cristallochimiques et physiques de phosphates condensés de terres rares. Paris: Publibook.

- Férid, M., Bernard, P. & Trabelsi-Ayedi, M. (1998). J. Therm. Anal. Calorim. 53, 227–234.
- Harms, K. & Wocadlo, S. (1995). XCAD4. University of Marburg, Germany. Horchani, K., Gâcon, J. C., Dujardin, C., Garnier, N., Garapon, C., Férid, M. &
- Ayadi, M. T. (2002). Nucl. Instrum. Methods Phys. Res. A, 486, 283–287. Horchani, K., Gâcon, J. C., Férid, M., Ayadi, M. T., Krachni, O. & Liu, G. K.
- (2003). Opt. Mater. 24, 169–174.
- Horchani-Naifer, K., Férid, M., Gâcon, J. C. & Trabelsi-Ayadi, M. (2006). Tunisian Patent SN06088.
- Makhov, V. N., Kirikova, N. Yu., Kirm, M., Krupa, J. C., Liblik, P., Lushchik, A., Lushchik, Ch., Negodin, E. & Zimmerer, G. (2002). Nucl. Instrum. Methods Phys. Res. A, 486, 437–442.
- Mazurak, Z., Lukowiak, E., Jezowska-Trzebiatowska, B., Schultze, D. & Waligora, Ch. (1984). J. Phys. Chem. Solids, 45, 487–493.
- Palkina, K. K., Maksimova, S. I. & Kuznetsov, V. G. (1978). Neorg. Mater. 14, 284–287. (In Russian.)
- Palkina, K. K., Saifuddinov, V. Z., Kuznetsov, V. G. & Chudinova, N. N. (1977). Sov. Phys. Dokl. 22, 698–700. (In Russian.)
- Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
- Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
- Wolinski, W., Malinowski, M. & Wolski, R. (1990). Proceedings of the 9th International Congress, Laser 89 Optoelektronik, Munich, 1989. Berlin: Springer.